Previous Year LPUNEST(B.TECH) Question Paper

Section - English

This section contains **25 Multiple Choice Questions**. Each question has four choices (A), (B), (C) and (D) out of which **ONLY ONE** is correct.

1. Choose the corr	ect article and fill in the	e blanks:	
How many star	s are there in	sky?	
A. No article	B. The	C. A	D. Big
2. Choose the corr	ect modal verb and fill	in the blanks:	
Mary	been to the superm	arket; the cupboards a	re all full.
A. Can			D. Must Have
3. Use the correct	form of adjective in the	e sentence below.	
All members of	f the family were at h	ome for the holidays.	What could make for
[Diwali than that?		
A. a happy	B. a happier	C. a more happier	D. a happiest
4. Identify the adje	ctival clause in the ser	ntence below.	
The trousers w	hich were gifted to me	by my father were qui	te expensive
A. The trousers	which were gifted		
B. The trousers	s which were gifted to r	me by my father	
C. Which were	gifted to me		
D. Which were	gifted to me by my fatl	her	
5. Select the corre	ct conjunction from the	e choices given to join	the sentence.
She cannot wa	lk properly. The doctor	r removes her plaster.	
A. as	B. until	C. when	D. although
6. Choose the sen	tence with 'Better' use	d as Noun	
A. My suit is be	etter than yours.		
B. We can alwa	ays learn something fro	om our betters.	
C. Out countryr	men can better their lo	t.	
D. You have do	one better by refusing t	to help him.	
7. "Myself" is a:			
A. Relative Pro	nouns	B. Demonstrative F	Pronouns
C. Reflexive Pr		D. Indefinite Prono	
8. The car was pa	rked in front of the st	ore. Choose the prepo	ositions that would be
	e one underlined.		
A. Across		C. On	D. At the back
9 It is dea		erjection which is not a	ppropriate for use.
A. Alas!	B. What a pity!	C. Oh!	D. Yay!
		the following sentence	
		in the parking met	
A. Work	B. Works	C. are working	D. were working
	sentences does not c		
A. The child rar	n happily towards his n	nother.	

	B. Brendan gently woke the sleeping ba	ıby.	
	C. Sali walked to the shops.		
	D. I visited my mum yesterday.		
12	. Fill in the blanks.		
	The circumference of a circle 3.14	1159265 times its dia	ameter no matter how
	small or large it is.		
	A. Measures	B. Measured	
	C. Will be measuring	D. Had been measi	ured
13	. Fill in the blanks.		
	While mom the VCD I hired, I	my assig	nment
	A. Watch/ would finish	B. Was going to wa	tch/ finishes
	C. Was watching/ finished	D. Will watch/ am fi	nishing
14	. Choose the correct option:		
	They have completed 24 years of together	etherness today and	by next year on the
	same day, they their 25th anniv	ersary.	
	A. Will have celebrated	B. Will be celebrating	ng
	C. Are celebrating	D. Have been celeb	orating
15	. Find out the synonyms of 'Defray'?		_
	A. Exit B. Spend	C. Malicious	D. Alight
16	. Which of the following options is an ant	onym for 'Destitute'?	
	A. Exhausted B. Impoverished	C. Affluent	D. Poor
17	. 'One who runs away from justice' provid	de one word for the e	expression.
	A. Extravagant B. Eccentric	C. Fugitive	D. Connoisseur
18	. Choose the correct meaning of the und	erlined phrase.	
	We are afraid that you may be led astra	<u>y</u> in Arvind's bad cor	npany.
	A. Misguided B. Lose the job	C. Killed	D. Get into trouble
19	. Which word is used for "Path of travel"?)	
	A. Course B. Coarse	C. Corse	D. Caerse
20	. Choose the appropriate homonym:		
	If one wants to reach God, he/she has t	o worldly t	things.
	A. Fergo B. Feorego	C. Forgeo	D. Forgo
21	. Read the sentence and Choose the Ind	lependent clause –	
	While you were at recess, we were eating	ng cake and ice crea	m.
	A. While you were at recess, we were e	ating cake and ice ci	ream.
	B. we were eating cake and		
	C. we were eating cake and ice cream		
	D. while you were at recess		
22	. What is true of the following example w	ith regards to subord	linate clause?
	The cat whom we are watching just cau	ght a mouse!	
	A. It contains a prepositional phrase.		
	B. It contains an adjectival clause.		
	C. It contains an adverbial clause.		
	D. It contains a noun clause.		

23. Read the following paragraphs and carefully determine what the main idea is for each.

There are no effective boundaries when it comes to pollutants. Studies have shown that toxic insecticides that have been banned in many countries are riding the wind from countries where they remain legal. Compounds such as DDT and toxaphene have been found in remote places like the Yukon and other Arctic regions.

This paragraph best supports the statement that

- A. toxic insecticides such as DDT have not been banned throughout the world.
- B. more pollutants find their way into polar climates than they do into warmer areas.
- C. studies have proven that many countries have ignored their own anti-pollution laws.
- D. DDT and toxaphene are the two most toxic insecticides in the world.
- 24. Spot the error in the sentence below. If no error, then you may choose the required option as well: -

None of two girls / who were present / appeared to be inclined / to listen to sane advice.

A. None of two girls

B. Who were present

C. Appeared to be inclined

D. No error

25. Choose some relationship from given four choices as given in original pair.

Teeth: Chew

A. Mind: Think

B. Sweater: Heat

C. Food: Taste

D. Eyes: flicker

Section – Chemistry

This section contains **25 Questions (20 Multiple Choice Questions and 5 Fill in the Blanks)**. Each **Multiple choice question** has four choices (A), (B), (C) and (D) out of which **ONLY ONE** is correct. For **Fill in the Blank** type question, enter the correct numerical value upto TWO decimal places.

1. The empirical formula of a compound is CH₂O. If 0.0835 moles of the compound contain 1 gm of hydrogen, the molecular formula of the compound is

A. $C_2H_4O_2$

B. C₃H₆O₃

C. $C_3H_6O_3$

 $D. C_3H_6O_3$

2. A solid has a structure in which W atoms are located at the corners of the unit cell, O atoms are located at the cube edge and Na atoms at the cube Centres. The formula of the compound is:

A. Na₂WO₃

B. NaWO₃

C. Na₂W₂O₄

D. Na₂WO₆

3. The incorrect electronic arrangement is

A. 2, 8, 13, 1

B. 2, 8, 12, 2

C. 2, 8, 8, 1

D. 2, 8, 8, 2

4. A molecule AX₂ has two lone pairs over A. Its shape is

A. Tetrahedral

B. Pyramidal

C. Angular

D. Linear

5. If PCl_5 is heated in two separate vess	sels of volume 5 lit and 10 lit respectively at
27°C . The extent of dissociation of P	Cl ₅ will be
A. More in 5 lit vessel	B. More in 10 lit vessel
C. More in 8 lit vessel	D. Cannot be sai
6. Milk is an example of	
A. Emulsion B. Suspension	C. gel D. true solution
7. Which of the following is the correct or	der of electron affinity? C. F > Cl > Br > I D. I < Br < F < Cl
8. Ellingham diagram represents	0.1 20120121 0.1 01 01 01
A. change of ΔG with temperature	B. change of ΔH with temperature
C. change of ΔG with pressure	D.change of $(\Delta G - T\Delta S)$ with temperature
9. Which of the following equations deno	otes that H ₂ O ₂ acts as a reducing agent?
A. PbS + $4H_2O_2 \rightarrow PbSO_4 + 4H_2O_4$	2 2
B. NaNO ₂ + H ₂ O ₂ \rightarrow NaNO ₃ + H ₂ O	
C. Ag ₂ O + H ₂ O ₂ \rightarrow 2Ag + O ₂ + H ₂ O	
D. 2KI+H ₂ O ₂ +H ₂ SO ₄ →I ₂ +K ₂ SO ₄ +	
10. Which of the following gives apple gre	_
A. Be B. Ca	C. Sr D. Ba
_	strength of oxy acids of group 15 element is
	$_{4}$ $_{\mathrm{B.}}$ $_{\mathrm{H_{3}PO_{4},H_{3}AsO_{4}H_{3}SbO_{4}}$, $_{\mathrm{HNO_{3}}}$
C. $HNO_{3,}H_{3}PO_{4,}H_{3}AsO_{4}H_{3}SbO_{4}$	$_{D.}$ $HNO_{3,H_3}PO_{4,H_3}AsO_{4}$ H_3SbO_{4}
The pair of Xenon compounds which on the central atom is	have same number of lone pairs of electrons
A. XeO_3 , XeF_6 B. XeF_2 , XeF_4	C. XeF_2 , XeO_3 D. XeF_4 , XeF_4
13. Bohr Magneton value in S.I. Units is	
A. $9.273 \times 10^{-24} \text{erg T}^{-1}$	B. $9.273 \times 10^{-24} \mathrm{J}\mathrm{T}^{-1}$
C. $9.273 \times 10^{-17} \mathrm{J}\mathrm{T}^{-1}$	D. $9.273 \times 10^{-10} \text{ cal T}^{-1}$
14. sp^3d^2 hybridisation is present in	
A. $\left[CoF_6\right]^{-3}$ B. $\left[Ni\left(CO\right)_4\right]$	C. $[Co(NH_3)_6]^{+2}$ D. All
15. Haemoglobin of the blood forms carb	oxy haemoglobin with
A. CO ₂ B. CO	$C. SO_2$ $D. NO_2$
16. Heterolytic fission of an organic coval	•
A. Free radicals	B. Both cation and anion
C. Only cation 17. Huckel's rule of aromaticity is	D. Only anion
A. having 6π electrons	B. having 3 double bonds
C. having (4n+2) π electrons	D. having alternate double bonds

 18. Which of the following order is A. Phenol > o - Cresol > o - N B. Phenol > o - Cresol < o - N C. Phenol < o - Cresol < o - D D. Phenol < o - Cresol > o - N 19. On ozonolysis 2-methyl-2-b 	Nitrophenol Nitrophenol Nitrophenol Nitrophenol	ing the acidic r	nature of phenol?	
A. 2moles of CH ₃ –CHO	B.	2molesof CH ₃	3COCH3	
C. $CH_3CHO \& CH_3COCH_3$	D.	CH ₃ CHO & H	CHO	
20. Aniline on heating with 'X' in t smell. Which of the following is A. CH ₃ Cl B. CHCl ₃	s 'X'?	_	a product with very ba D. C ₂ H ₅ Cl	ıd
			_ 0	
21. For the following question, enter the numerical value has more than two (For example: Numeric value 5 will be The maximum oxidation state)	decimal places, oe written as 5.0	, round-off the va 00 and 2.346 will	lue to TWO decimal places	
22. For the following question, enter the numerical value has more than two (For example: Numeric value 5 will be The van't Hoff factor for 0.7 dissociation of Barium nitrate is	decimal places, be written as 5.0 1M Barium	, round-off the va 00 and 2.346 will	lue to TWO decimal places be written as 2.35)	
23. For the following question, enter the numerical value has more than two (For example: Numeric value 5 will be	he correct num decimal places, pe written as 5.0	, round-off the va 00 and 2.346 will	lue to TWO decimal places be written as 2.35)	•
$E^0_{Zn^2/Zn} = -0.76V$ The EMF σ	of the cell	$Zn/Zn_{(IM)}^{2+} \parallel F$	$HCl(pH=2) H_{2(1atm)}P$	ľ
is V.				
24. For the following question, enter the numerical value has more than two (For example: Numeric value 5 will be 75% of a first order reaction is have been completed in	decimal places, be written as 5.0 completed in	, round-off the va 00 and 2.346 will	lue to TWO decimal places be written as 2.35)	
25. For the following question, enter the numerical value has more than two (For example: Numeric value 5 will be 0.2 g of an organic compound	he correct num decimal places, pe written as 5.0	, round-off the va 00 and 2.346 will	lue to TWO decimal places be written as 2.35)	•
then the percentage of carbon	in it is	·		

Section – Mathematics

This section contains 25 Questions (20 Multiple Choice Questions and 5 Fill in the Blanks). Each Multiple choice question has four choices (A) (B) (C) and (D) out of which ONLY ONE is

cor	rect. For Fill in the Blank type quees.	, , , ,	, , , ,	
	1. Universal set, $U = \{x / x^5 - 6x^4 + 11x^3 - $	$Sx^2 = 0$		
	A. {1, 3}	B. {1, 2, 3}	C. {0, 1, 3}	D. {0, 1, 2, 3}
	2. If R is an equivalence relati	on on a set A, then F	R ^{−1} is	
	A. reflexive only		B. symmetric but no	ot transitive
	C. equivalence		D. None of the abo	ve
	3. Lt $\propto \left(\frac{x^2+5x+3}{x^2+x+2}\right)^x$			
	A. e ⁴	B. e ²	C. e ³	D. e
	4. If one root of the quadratic	equation ax2+bx+c=0) is 3-4i then 31a+b-	⊦c=
	A. 0	B. 2a	C. 2b	D. 2c
	5. If the 2 nd , 5 th and 9 th terms ratio of this G.P.is:	of a non-constant A	A.P. are in G.P., the	n the common
	A. 8/5	B. 4/3	C. 1	D. 7/4
	6. If $A(\alpha) = \begin{pmatrix} \cos \alpha & \sin \alpha \\ -\sin \alpha & \cos \alpha \end{pmatrix}$ then $A(\alpha) = \begin{pmatrix} \cos \alpha & \sin \alpha \\ \cos \alpha & \cos \alpha \end{pmatrix}$	$A(\alpha) A(\beta) =$		
	A. $A(\alpha)+A(\beta)$ 7. A square matrix (Nonsingular)	B. $A(\alpha)$ - $A(\beta)$ lar) satisfies $A^2 - A$ +		D. A(α-β)
	A. $\frac{I-A}{2}$	B. I-A	C. $\frac{I+A}{2}$	D. I+A
	8. The ratio in which $\overline{i} + 2\overline{j} + 3$ A. 1 : 2		$-2\overline{i} + 3\overline{j} + 5\overline{k}$ and C. 3:4	
	9. Consider the following statemP: Suman is brilliantQ: Suman is richR: Suman is honest	ents		
	The negation of the statemerich" can be expressed as:	ent "Suman is brilliant	and dishonest if and o	only if Suman is
	A. \sim (P Λ \sim R) \leftrightarrow Q	B. $\sim P \Lambda(Q \leftrightarrow \sim R)$	C. \sim (Q \leftrightarrow (P $\land \sim$ R)	D. $\sim Q \leftrightarrow \sim P \wedge R$
	10. 10 men and 6 women are	to be seated in a row	so that no two wom	en sit together.
	The number of ways they	can be seated is		
	A. 11!10!	B. (11! /6!5!)	C. (10!9! /5!)	D. (11!10! /5!)
	11. $1^3 - 2^3 + 3^3 - 4^3 + \dots + 9$	³ =		

	A. 425	B425	C. 475	D475
12.	The ratio of the coefficient	of x^{15} to the term in	ndependent of x in the	e expansion
	of $\left(x^2 + \frac{2}{x}\right)^{15}$ is			
	A. 1:32	B. 1:4	C. 7:16	D. 7:64
13.	Let f be a polynomial funct	tion such that f(3x) =		Then:
	A. $f^{11}(2) - f^{1}(2) = 0$		B. $f^{11}(2) - f(2) = 4$	
11	C. $f(2) - f^{1}(2) + f^{11}(2) = 10$	a paint on the aurus	D. $f(2) + f^{1}(2) = 28$	n io
14.	The minimum distance of a A. $(\sqrt{15})/2$			n is D. (√19)/2
	• •	<i>D.</i> ((10/2)	J. 1(10/2)	D. (110)/2
15.	$\int \frac{dx}{\cos x + \sqrt{3}\sin x} \text{ equals}$			
	A. $\log \tan \left(\frac{x}{2} + \frac{\pi}{12} \right) + c$		B. $\log \tan \left(\frac{x}{2} - \frac{\pi}{12} \right) +$	c
	C. $\frac{1}{2} \log \tan \left(\frac{x}{2} + \frac{\pi}{12} \right) + c$		D. $\frac{1}{2}\log\tan\left(\frac{x}{2} - \frac{\pi}{12}\right)$	+ <i>c</i>
16.	$\int \frac{1}{(x+100)\sqrt{x+99}} dx = f(x)$	+c then f(x) =	, , , , ,	
	A. $2(x+100)^{1/2}$		B. 3(x+100) ^{1/2}	
	C. $2\tan^{-1}\sqrt{x+99}$		D. $2\tan^{-1}\sqrt{x+100}$	
			V 11 100	
17.	$\int_{0}^{\pi} \frac{x \operatorname{Sinx}}{1 + \operatorname{Cos}^{2} x} dx =$			
	Α. π ² /4	B. π ² /2	C. π ² /3	D. π^2
18.	If $y = y(x)$ is the solution	of the differential	equation, $x\frac{dy}{dx} + 2y =$	$= x^2$ satisfying
	$y(1) = 1$, then $y\left(\frac{1}{2}\right)$ is equal	al to :		
	A. 1/4	B. (7/64)	C. (49/16)	D. (13/16)
19.	If the mid points of the sid		_	re respectively
	(2, 1), (-1, -2) and (3, 3), th			_
	A. x-2y=0	B. 5x-4y=6	C. 2x+3y=8	D. 3x-2y=6
20.	Two vertices of a triangle this triangle, then the coord	dinates of the third ve	-	
	A. (4,7)	$B.\left(-2,\frac{-7}{2}\right)$	C. (-4,-7)	D. (-2,3)
21.	For the following question, en	ter the correct numeric	al value upto TWO decir	mal places. If the
	numerical value has more than	two decimal places, rou	nd-off the value to TWO	decimal places.
	(For example: Numeric value 5 will be written as 5.00 and 2.346 will be written as 2.35)			

If z, iz and z+iz are the vertices of a triangle and if |z| = 4, then the area (in sq. units) of that triangle, is _____.

- 22. For the following question, enter the correct numerical value upto TWO decimal places. If the numerical value has more than two decimal places, round-off the value to TWO decimal places. (For example: Numeric value 5 will be written as 5.00 and 2.346 will be written as 2.35)

 There are 10 points in a plane out of which 6 are collinear. The number of straight lines formed by joining all these points is ______.
- 23. For the following question, enter the correct numerical value upto TWO decimal places. If the numerical value has more than two decimal places, round-off the value to TWO decimal places. (For example: Numeric value 5 will be written as 5.00 and 2.346 will be written as 2.35)

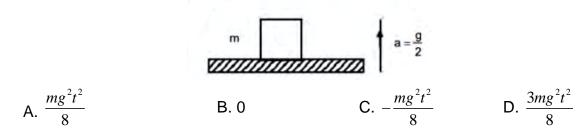
 Minimum number of times a fair coin must be tossed so that the probability of getting at least one head is more than 99% is ______.
- 24. For the following question, enter the correct numerical value upto TWO decimal places. If the numerical value has more than two decimal places, round-off the value to TWO decimal places. (For example: Numeric value 5 will be written as 5.00 and 2.346 will be written as 2.35) An envelope is known to have come from either 'LONDON' OR 'CLIFTON'. On the postal card only two successive letters ON are visible. The probability that the envelope comes from LONDON is 12/____.
- 25. For the following question, enter the correct numerical value upto TWO decimal places. If the numerical value has more than two decimal places, round-off the value to TWO decimal places. (For example: Numeric value 5 will be written as 5.00 and 2.346 will be written as 2.35)

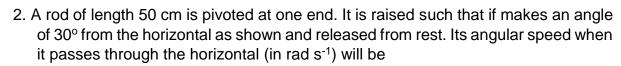
The eccentricity of the ellipse
$$\frac{x^2}{25} + \frac{y^2}{16} = 1$$
 is $3/$ ___.

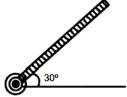
Section - Physics

This section contains **25 Questions (20 Multiple Choice Questions and 5 Fill in the Blanks)**. Each **Multiple choice question** has four choices (A), (B), (C) and (D) out of which **ONLY ONE** is correct. For **Fill in the Blank** type question, enter the correct numerical value upto TWO decimal places.

1. A block of mass m is kept on a platform which starts from rest with constant acceleration g/2 upward, as shown in figure. Work done by normal reaction on block in time t is:







A. √30	B. (√20)/3	C. (√30)/2	D. (√40)/2
/ \. \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\	D. (\20)/0	O. (100)/2	D. (\\\\)/2

3. A long cylindrical rod is welded to a thin circular disc of diameter 0.5 m at a point on its circumference. The rod is in the same plane as that of the disc and forms a tangent to the disc. The radius of gyration of the disc about the rod (in m) is

A. 1/4 B. $(\sqrt{5})/8$ C. 1/2 D. $2\sqrt{2}$

4. Expression for time in terms of G (universal gravitational constant), h (Planck constant) and c (speed of light) is proportional to:

A.
$$\sqrt{\frac{hc^5}{G}}$$
 B. $\sqrt{\frac{Gh}{c^3}}$ C. $\sqrt{\frac{c^3}{Gh}}$

5. In a car race on straight road, car A takes a time 't' less than car B at the finish and passes finishing point with a speed 'v' more that of car B. Both the cars start from rest and travel with constant acceleration a₁ and a₂ respectively. Then 'v' is equal

A.
$$\frac{2a_1a_2}{a_1 + a_2}t$$
 B. $\sqrt{2a_1a_2}t$ C. $\frac{a_1 + a_2}{2}t$ D. $\sqrt{a_1a_2}t$

6. A shell is fired from a fixed artillery gun with an initial speed u such that it hits the target on the ground at a distance R from it. If t₁ and t₂ are the values of the time taken by it to hit the target in two possible ways, the product t₁t₂ is:

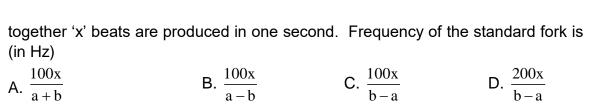
7. Moon is revolving in a circular orbit of radius 60R (R = radius of earth). Assume that the radius of the moon is R/4. If the moon is stopped for an instant and then released, it will fall towards the earth. Ignoring the atmospheric friction, the velocity of moon just before it strikes the earth is (Take g = acceleration due to gravity at the surface of earth)

A.
$$\sqrt{\frac{59}{30}} gR$$
 B. $\sqrt{\frac{118}{61}} gR$ C. $\sqrt{\frac{7gR}{10}}$ D. $\sqrt{\frac{47}{30}} gR$

8. A particle executing SHM along a straight line has zero velocity at points A and B whose distances from 'O' on the same direction OAB are 'a' and 'b' respectively. If the velocity at the midpoint between A and B is 'v', then its time period is

A.
$$\frac{\pi(b+a)}{v}$$
 B. $\pi\left(\frac{b-a}{v}\right)$ C. $\left(\frac{b+a}{2v}\right)$

9. The frequency of a tuning fork P is 'a' % less than a standard fork A. The frequency of another fork Q is 'b' % greater than that of A. When P and Q are sounded



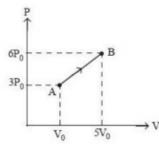
10. A tank full of water has a small hole at the bottom. If one-fourth of the tank is emptied in t₁ seconds and the remaining three-fourths of the tank is emptied in t₂ seconds. Then the ratio t₁/t₂ is



11. A metallic wire of density d floats horizontal in water. The maximum radius of the wire so that the wire may not sink, will be (surface tension of water = T)

A.
$$\sqrt{\frac{2T}{p\,dg}}$$
 B. $\sqrt{\frac{2p\,T}{dg}}$ C. $\sqrt{\frac{2p\,Tg}{d}}$ D. $\sqrt{2p\,Tgd}$

12. One mole of a monoatomic ideal gas undergoes the process A→B in the given P-V. Diagram Specific heat capacity in the process is



13. An ideal gas enclosed in a cylinder at pressure of 2 atm and temperature. 300K. The mean time between two successive collisions is 6x10⁻⁸ s. If the pressure is doubled and temperature is increased to 500K the mean time between two successive collisions will be close to:

A. 3x10⁻⁶ s

B. 4x10⁻⁸ s

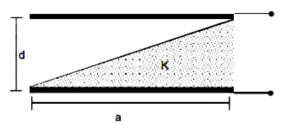
C. 2x10⁻⁷ s

D. 5x10⁻⁸ s

14. Two point charges
$$q_1(\sqrt{10}\mu C)$$
 and $q_2(-25\mu C)$ are placed on the x-axis at $x=1m$ and $x=4m$ respectively. The electric field (in V/m) at a point $y=3m$ on y-axis is,

$$\begin{bmatrix} \text{take} - \frac{1}{4\pi g \varepsilon_0} - 9 \times 10^9 \, \text{Nm}^2 \text{C}^{-2} \end{bmatrix}$$
A. $(-81\hat{\imath} + 81\hat{\jmath}) \times 10^2$
B. $(81\hat{\imath} + 81\hat{\jmath}) \times 10^2$
C. $(-63\hat{\imath} + 27\hat{\imath}) \times 10^2$
D. $(63\hat{\imath} + 27\hat{\imath}) \times 10^2$

15. A parallel plate capacitor is made of two square plates of side 'a', separated by a distance d (d << a). The lower triangular portion is filled with a dielectric of dielectric constant K, as shown in the figure., Capacitance of this capacitor is:



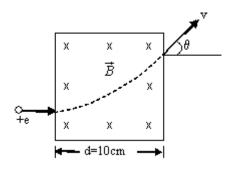
$$A. \frac{K \in_0 a^2}{2d(K+1)}$$

C.
$$\frac{K \in_0 a^2}{d(K-1)} \ell nK$$

B.
$$\frac{K \in_0 a^2}{d} \ell n K$$

$$\mathsf{D.}\ \frac{1}{2}\frac{K\in_0 a^2}{d}\ell nK$$

16. A proton accelerated by a pd V=500 KV moves through a transverse magnetic field B=0.51 T as shown in figure. Then the angle θ through which the proton deviates from the initial direction of its motion is (approximately)



A. 15°

- B. 30°
- C. 45°
- D. 60°

- 17. A magnetic dipole in a constant magnetic field has
 - A. zero potential energy when the torque is maximum
 - B. minimum potential energy when the torque is maximum
 - C. maximum potential energy when the torque is maximum
 - D. zero potential energy when the torque is minimum
- 18. If λ_1 and λ_2 are the wavelength of the photons emitted when electrons in nth orbit of hydrogen atom fall to first excited state and ground state respectively, then the value of n is

A.
$$\sqrt{\frac{2(\lambda_2 - \lambda_1)}{2\lambda_2 - \lambda_1}}$$

C.
$$\sqrt{\frac{4\lambda_2 - \lambda_1}{4(\lambda_2 - \lambda_1)}}$$

B.
$$\frac{2\lambda_2 - \lambda_1}{2(\lambda_2 - \lambda_1)}$$

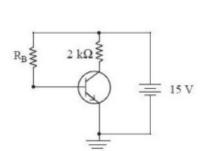
D.
$$\sqrt{\frac{4(\lambda_2 - \lambda_1)}{4\lambda_2 - \lambda_1}}$$

- 19. The energy released per fission of $_{92}U^{235}$ is 200 Mev. The fission rate of $_{92}U^{235}$ required to produce 2 watt power is
 - A. 1.25 x 10²⁶ per second

B. 2.56 x 10²⁶ per second

C. 1.25 x 10¹³ per second

- D. 6.25 x 10¹⁰ per second
- 20. In the following common emitter circuit, β = 100 and V_{CE} = 7V. If V_{BE} is negligible, then the base current is



	_		
A. 0.015 mA	B. 0.045 mA	C. 0.025 mA	D. 0.035 mA
	ng question, enter the correct nu has more than two decimal place	•	•
(For example: No	umeric value 5 will be written as 5	5.00 and 2.346 will be writt	en as 2.35)
A 1 kW carrie wave is	r is modulated to a depth ofKW.	80%. The total power	in the modulated
22. For the following	ng question, enter the correct nu	merical value upto TWO d	ecimal places. If the
	has more than two decimal place umeric value 5 will be written as 5		•
A mass of 10k	kg is suspended vertically by	a rope form the roof. V	Vhen a horizontal
• • • • • • • • • • • • • • • • • • • •	ed on the rope at some point	•	•
•	he suspended mass is at 6 N (g=10ms ⁻²)	equilibrium, the magni	tude of the force
numerical value	has more than two decimal place	s, round-off the value to T	WO decimal places.
	umeric value 5 will be written as 5		· ·
left by 10 cm.	ging the resistances, the bal The resistance of their serie on the left slot before interc	es combination is 1k W	/. How much was
	ng question, enter the correct nu		
numerical value	has more than two decimal place umeric value 5 will be written as 5	s, round-off the value to T	WO decimal places.
	an object placed in front of a		
formed at a po	oint which is 10 cm more dis of the image is		•
•	ng question, enter the correct nu	merical value upto TWO d	ecimal places. If the
numerical value	has more than two decimal place umeric value 5 will be written as 5	s, round-off the value to T	WO decimal places.
A body moves	s along a circular path of rac ould be its angular velocity ir	lius 10m and the coeff	icient of friction is
	=9.8m/s ²)		Tom the dander

Section - Biology

This section contains **25 Multiple Choice Questions**. Each question has four choices (A), (B), (C) and (D) out of which **ONLY ONE** is correct.

1. During the propagation of a nerve impu	ulse, the action potential results from the
movement of A. Sodium ions from ECF to ICF	B. Sodium ions from ICF to ECF
C. Potassium ions from ECF to ICF	D.Potassium ions from ICF to ECF
2. Acromegaly is the result of	
A. Hypersecretion of GH in children	
B. Hypersecretion of GH in adults	
C. Hypersecretion of GH	
D. Deficiency of vitamin D	alandar of DO II are first accounted by
3. Electrons from the excited chlorophyll mo	
A. Pheophytin	B. Ferredoxin
C. Cytochrome f4. During which stage in the complete oxida	D. Cytochrome b
ATP molecules formed from ADP?	ation of glucose are the greatest number of
A. Kreb's cycle	
B. Glycolysis	
C. Electron transport chain	
D. Conversion of pyruvic acid to acetyl	CoA
5. Which of these bioactive compounds is	
organ-transplant patients?	
2 3 -	
A. Acetic acid B. Ethanol	C. Cyclosporine A D. Pectinase
A. Acetic acidB. EthanolWhich is not an example of transmembra	·
A. Acetic acidB. EthanolWhich is not an example of transmembra compartments?	ane transport between different subcellular
A. Acetic acidB. EthanolWhich is not an example of transmembra compartments?A. Transport from the stroma into thyla	ane transport between different subcellular koid space
A. Acetic acidB. EthanolWhich is not an example of transmembra compartments?A. Transport from the stroma into thylaB. Transport from the cytoplasm into the	koid space le lumen of the endoplasmic reticulum
 A. Acetic acid B. Ethanol Which is not an example of transmembra compartments? A. Transport from the stroma into thylal B. Transport from the cytoplasm into the C. Transport from the endoplasmic retion 	koid space le lumen of the endoplasmic reticulum culum into the Golgi complex
 A. Acetic acid B. Ethanol Which is not an example of transmembra compartments? A. Transport from the stroma into thylal B. Transport from the cytoplasm into the C. Transport from the endoplasmic retion D. Transport from mitochondrial intermatics 	koid space le lumen of the endoplasmic reticulum
 A. Acetic acid B. Ethanol Which is not an example of transmembra compartments? A. Transport from the stroma into thylal B. Transport from the cytoplasm into the C. Transport from the endoplasmic retion D. Transport from mitochondrial intermatrix 	koid space te lumen of the endoplasmic reticulum culum into the Golgi complex membrane space into the mitochondrial
 A. Acetic acid B. Ethanol 6. Which is not an example of transmembra compartments? A. Transport from the stroma into thyla B. Transport from the cytoplasm into the C. Transport from the endoplasmic retion D. Transport from mitochondrial intermatrix 7. The two amino acids having R groups with the company of the c	koid space le lumen of the endoplasmic reticulum culum into the Golgi complex membrane space into the mitochondrial ith a negative net charge at pH 7.0 are
 A. Acetic acid B. Ethanol 6. Which is not an example of transmembra compartments? A. Transport from the stroma into thylal B. Transport from the cytoplasm into the C. Transport from the endoplasmic retion D. Transport from mitochondrial intermatrix 7. The two amino acids having R groups with A. Aspartate and glutamate 	koid space le lumen of the endoplasmic reticulum culum into the Golgi complex membrane space into the mitochondrial ith a negative net charge at pH 7.0 are B. Arginine and histidine
 A. Acetic acid B. Ethanol 6. Which is not an example of transmembra compartments? A. Transport from the stroma into thylal B. Transport from the cytoplasm into the C. Transport from the endoplasmic retion. D. Transport from mitochondrial intermatrix 7. The two amino acids having R groups with A. Aspartate and glutamate C. Cysteine and methionine 	koid space le lumen of the endoplasmic reticulum culum into the Golgi complex membrane space into the mitochondrial ith a negative net charge at pH 7.0 are B. Arginine and histidine D. Proline and valine
 A. Acetic acid B. Ethanol 6. Which is not an example of transmembra compartments? A. Transport from the stroma into thylal B. Transport from the cytoplasm into the C. Transport from the endoplasmic retion D. Transport from mitochondrial intermatrix 7. The two amino acids having R groups with A. Aspartate and glutamate 	koid space le lumen of the endoplasmic reticulum culum into the Golgi complex membrane space into the mitochondrial lith a negative net charge at pH 7.0 are B. Arginine and histidine D. Proline and valine olexes of mitochondria that are involved in
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 A. Acetic acid B. Ethanol 6. Which is not an example of transmembra compartments? A. Transport from the stroma into thylal B. Transport from the cytoplasm into the C. Transport from the endoplasmic retic D. Transport from mitochondrial intermatrix 7. The two amino acids having R groups where A. Aspartate and glutamate C. Cysteine and methionine 8. The major electron transport chain comparts the generation of Reactive oxygen specific 	koid space le lumen of the endoplasmic reticulum culum into the Golgi complex membrane space into the mitochondrial ith a negative net charge at pH 7.0 are B. Arginine and histidine D. Proline and valine olexes of mitochondria that are involved in cies (ROS)
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 A. Acetic acid B. Ethanol 6. Which is not an example of transmembra compartments? A. Transport from the stroma into thylat B. Transport from the cytoplasm into the C. Transport from the endoplasmic retion D. Transport from mitochondrial intermatrix 7. The two amino acids having R groups where A. Aspartate and glutamate C. Cysteine and methionine 8. The major electron transport chain compart the generation of Reactive oxygen special complex I and complex IV C. complex I and complex IV 9. Name the state where never dividing cell A. G0 B. G1 	koid space le lumen of the endoplasmic reticulum culum into the Golgi complex membrane space into the mitochondrial lith a negative net charge at pH 7.0 are B. Arginine and histidine D. Proline and valine blexes of mitochondria that are involved in cies (ROS) B. complex I and complex III D. None of these s of neurons and skeletal muscle present? C. G2 D. M
 A. Acetic acid B. Ethanol 6. Which is not an example of transmembra compartments? A. Transport from the stroma into thylal B. Transport from the cytoplasm into the C. Transport from the endoplasmic retion. D. Transport from mitochondrial intermatrix 7. The two amino acids having R groups where A. Aspartate and glutamate C. Cysteine and methionine 8. The major electron transport chain compart the generation of Reactive oxygen special complex I and complex IV C. complex I and complex IV 9. Name the state where never dividing cell A. GO B. G1 10. One of the following is the correct sequence.	koid space le lumen of the endoplasmic reticulum culum into the Golgi complex membrane space into the mitochondrial lith a negative net charge at pH 7.0 are B. Arginine and histidine D. Proline and valine blexes of mitochondria that are involved in cies (ROS) B. complex I and complex III D. None of these s of neurons and skeletal muscle present? C. G2 D. M ence to make a transgenic animal.
 A. Acetic acid B. Ethanol 6. Which is not an example of transmembra compartments? A. Transport from the stroma into thylal B. Transport from the cytoplasm into the C. Transport from the endoplasmic retion. D. Transport from mitochondrial intermatrix 7. The two amino acids having R groups where A. Aspartate and glutamate C. Cysteine and methionine 8. The major electron transport chain compart the generation of Reactive oxygen special complex I and complex IV C. complex I and complex IV 9. Name the state where never dividing cell A. GO B. G1 10. One of the following is the correct sequence.	koid space le lumen of the endoplasmic reticulum culum into the Golgi complex membrane space into the mitochondrial lith a negative net charge at pH 7.0 are B. Arginine and histidine D. Proline and valine blexes of mitochondria that are involved in cies (ROS) B. complex I and complex III D. None of these s of neurons and skeletal muscle present? C. G2 D. M

B. Micro injection – transfection – elec	tro portion – retrovi	ral vectors – transomics
C. Transfection – micro injection – tran	•	
D. None of these	•	
11. In which organ of digestive tract hydrol	ysis of starch into r	naltose take place?
A. Stomach B. Liver	C. Mouth	D. Duodenum
12. The alveoli of the lungs do not contain	"air" because	
 A. We normally do not ventilate our lur 	ngs at a high enoug	ıh rate.
B. The lungs have too many alveoli to	ventilate.	
C. There is "dead space" in the tracher		
D. The trachea and bronchi are too sm		_
13. Which of the following statement about	t plasmid is correct	?
I) It can replicate by itself		
II) Hybrid plasmid is introduced into bac		
III) Other than bacteria, plasmid can als		
,	C. II and III only	•
14. Which one of the following is the corre menstrual cycle?	ct matering of the	events occurring during
A. Menstruation: breakdown of myome	etrium and ovum no	nt fertilized
B. Ovulation: LH and FSH attain pea		
progesterone	ar iovor and onarp	
C. Proliferative phase: Rapid regene	ration of myometr	ium and maturation of
Graafian follicle	,	
D. Development of corpus luteum: Se	cretory phase and	increased secretion of
progesterone		
15. Out of the total sunlight energy reachi	ing the atmosphere	e, the fraction utilized in
photosynthesis is approximately:		
A. 0.002% B. 0.02%	C. 0.2%	D. 2.0%
16. Depolarization of the T-tubule membra	ne activates the sa	rcoplasmic reticulum via
the:		
A. Ryanodine receptor		
 B. Dihydropyridine receptors C. Increased Na⁺ and K⁺ conductance 	in and plata mamb	arano
D. IP ₃ receptor	in end-plate memi	nane
17. Consider the following statement:		
a. The banded appearance of the Sard	comere is due to di	fference in the size and
density of thick and thin filaments		
b. The A band is the area containing th	ick filaments	
c. A band includes the M line, the H ba		f overlap (thick and thin
filaments)		
 d. A band and I band are anisotropic ar 	nd isotropic, respec	tively
The incorrect statements are:		
A. a, b and c B. b and c	C. b, c and	D. None

18. A sedentary sea anemone gets attach association is	ned to the shell lining of hermit crab. The
	D. Cymhinnia
A. Ectoparasitism	B. Symbiosis
C. Commensalism	D. Amensalism
The zone at the edge of a lake or ocea immersed in water is called	n which is alternatively exposed to air and
A. pelagic zone	B. benthic zone
C. lentic zone	D. littoral zone
20. Elicitors are molecules that	
A. Induce cell division	
B. Stimulate production secondary met	abolites
C. Stimulate hairy root formation that a	
D. None of these	
21. Rheumatoid arthritis is different from se	ome other forms of arthritis as it
A. Generally, occurs above the waist	ome carer rome or aramae as a
B. Is more painful than other forms	
C. Is symmetrical, affecting the right an	nd the left sides of the hody
D. Occurs below the waist	a the left sides of the body
22. IUCN (The International Union for Cons	ervation of Nature and Natural Resources)
headquarters is at	ervation of Mature and Matural Nesources)
A. Morges, Switzerland	B. Paris, France
<u> </u>	
C. Vienna, Austria	D. New York, USA
23. Which of the following organisms for pollution?	ound in numan waste that cause water
A. Coliform bacteria	B. Viruses
C. Protozoa	D. Parasitic worms
24. What is tautonym?	
A. These are the repeated sequence	
B. It is a name of fish	
C. Identical name of genus and species	3
D. It is a name of the genus	
25. T.O. Diener discovered	
A. Bacteriophage	B. Infectious protein
C. Free infectious DNA	D. Free infectious RNA
C. 7 TOO HIROGIOGO DIWY	5.1130 III000000 1000